

14PHY12/22

First/Second Semester B.E. Degree Examination, Dec.2015/Jan.2016 Engineering Physics

Time: 3 hrs.

Max. Marks:100

Note: 1. Answer any FIVE full questions, selecting atleast ONE full question from each part.
2. Physical constants: Velocity of light, C = 3 × 10⁸ m/s; Plank's constant, h = 6.625 × 10⁻³⁴ J.S; Mass of electrons, m = 9.11 × 10⁻³¹ kg; Boltzmann's constant, K = 1.38 × 10⁻²³ J/K. Avogadro number, N_A = 6.02 × 10²⁶/K mole.

PART-1

- a. Define phase velocity and group velocity. Derive a relation between the two. (05 Marks)
 - b. What is the physical interpretation of wave function? Explain the nature of eigen values and eigen functions.

 (06 Marks)
 - c. Explain Wein's law and Rayleigh Jean's law Discuss their drawbacks. (06 Marks)
 - d. Calculate the de Broglie wavelength associated with an electron carrying energy 2000 eV.
- 2 a. State Heisenberg's uncertainity principle. Using uncertainity principle. Explain the non existence of electron in the nucleus. (07 Marks)
 - b. Using time independent Schrodinger's wave equation, obtain the expression for the normalized wave function for a particle in one dimensional potential well of infinite height.
 - c. The speed of electron is measured to within an uncertainty of 2.2 × 10⁴ m/s in one dimension. What is the minimum width required by the electron to be confined in an atom? (04 Marks)

PART - 2

- a. Explain the probability of occupation of various energy state by electron at T = 0 K and T > 0 K on the basis of Fermi factor. (06 Marks)
 - b. Define Hall Effect and Hall Voltage. Derive an expression for Hall coefficient. (06 Marks)
 - c. Explain BCS theory of Super conductivity. (04 Marks)
 - d. Find the relaxation time of conduction electrons in a metal of resistivity $1.54 \times 10^{-8} \Omega \text{m}$, if the metal has $5.8 \times 10^{28} \text{ electrons/m}^3$. (04 Marks)
- 4 a. Discuss different types of super conductors. (04 Marks)
 - b. Explain Fermi energy and Fermi factor. (06 Marks)
 - c. Explain failure of Classical free electron theory. (06 Marks)
 - d. Calculate the Fermi velocity for the free electrons in gold. Given $E_F = 5.53 \text{eV}$. (04 Marks)



PART-3

- 5 a. Derive an expression for energy density in terms of Einstein's coefficients. (08 Marks)
 - b. Explain the construction and working of carbon dioxide laser device. (08 Marks)
 - c. The attenuation of light in an optical fiber is 3.6 dB/km. What fraction of its initial intensity remains after i) 1 km ii) after 3 km. (04 Marks)
- 6 a. What is Total internal reflection? Derive an expression for acceptance angle of an optical fiber.

 (08 Marks)
 - b. Discuss different types of optical fibres.

(06 Marks)

c. An optical fiber has a numerical aperture of 0.32. The refractive index of cladding is 1.48.
 Calculate the refractive index of the core, the acceptance angle of the fiber and the fractional index change.

PART - 4

- 7 a. Obtain the expression for inter planar spacing of a cubic crystal. (05 Marks)
 - b. Calculate the atomic packing factor for SC, FCC and BCC lattices. (06 Marks)
 - c. Write a note on Perovskite structure. (06 Marks)
 - d. A sodium chloride crystal is used as a diffraction grating with X rays. For the d_{111} spacing of the chloride ions the angle of diffraction 2θ is 27.5° . If the lattice constant of the crystal is 0.563nm, what is the wavelength of X rays? (03 Marks)
- 8 a. What is Bragg's law? Explain how Bragg's spectrometer is used for determination of interplanar spacing in a crystal. (08 Marks)
 - b. Discuss the principle and working of Liquid Crystal Display. (08 Marks)
 - c. Draw (100), (110), (011) and (111) planes in a Simple cubic crystal. (04 Marks)

PART-5

- 9 a. Distinguish acoustic, subsonic and supersonic waves. (04 Marks)
 - b. Explain the preparation of nano structure using Sol Gel method. (06 Marks)
 - c. Write a note on Carbon Nanotubes. (06 Marks)
 - d. What are Shock waves? Mention few applications of Shock wave. (04 Marks)
- a. Explain the principle, construction and working of Reddy Shock tube. (08 Marks)
 - b. Explain the preparation of nano structures using Top Down approach method. Mention any two properties of nano materials. (06 Marks)
 - c. Explain the construction and working of Scanning Electron Microscope. (06 Marks)